

**COMMON ELECTIVE GENERIC / DISCIPLINE SPECIFIC  
FROM THE ACADEMIC YEAR 2023-24**

<b>Sl. No</b>	<b>Sem</b>	<b>Title of the Paper</b>
1.	III	PRINCIPLES OF ECONOMICS I
2.	IV	PRINCIPLES OF ECONOMICS II
3.	III	PHYSICS – I
4.	IV	PHYSICS – II PHYSICS PRACTICALS – II
5.	III	CHEMISTRY – I
6.	IV	CHEMISTRY – II CHEMISTRY PRACTICALS – II
7.	III	DISCRETE MATHEMATICS I
8.	IV	DISCRETE MATHEMATICS II
9.	III	STATISTICAL METHODS & THEIR APPLICATIONS-I
10.	IV	STATISTICAL METHODS & THEIR APPLICATIONS-II STATISTICAL METHODS & THEIR APPLICATIONS PRACTICALS –II
11.	III	MICROBIOLOGY I
12.	IV	MICROBIOLOGY II MICROBIOLOGY PRACTICALS II

## PRINCIPLES OF ECONOMICS I

### Course Objectives:

To provide a frame work of knowledge relating to the concepts and practice of Economics in Indian context and to make the students understand the application of Economic principles in the strategic sector. Also, to provide insight on the most pressing issue “Demand for Defence Expenditure” i.e. the right size of Defence Budget.

### Expected Course Outcomes:

On the successful completion of the course, student will be able to:

1	To learn about an introduction to the discipline of Economics.	K1
2	Understand about the concepts used in Economics	K2
3	Identify the various Mechanisms in economics.	K2
4	Acquire knowledge about importance of market forms	K3
5	To know about economics related to Defence	K4

**K1** - Remember; **K2** - Understand; **K3** - Apply; **K4** - Analyze; **K5** - Evaluate; **K6** - Create

### Unit:1

Economic Analysis - Basic Problems of Economy - Economic Systems - Capitalism - Socialism - Mixed Economy - Communalism - Role of Government.

### Unit:2

Concept and Management of National Income - Problems of Measurement - Trends in National Income and Planning.

### Unit:3

Market Mechanism - Law of Demand and Supply - Elasticity of Demand - Elasticity Measurement - Uses - Limitations.

<b>Unit:4</b>		
Market Forms - Perfect Competition - Monopoly - Discriminative Monopoly - Monopolistic Competition - Wastes of Monopolistic Competition.		
<b>Unit:5</b>		
Defence Economics - Economics of Conflict and Terrorism - Scope and Definition - Micro and Macro Economic impact – International Monetary Organisation – IMF, IBRD, WTO, GATT, TRIPS and TRIMS.		
<b>Reference</b>		
1	Dr. S. Sankaran, Micro Economics.	
2	M.L. Jhingan, Micro Economics	
3	Dutt and Sundaram, Indian Economy.	
4	Defence Economics,	

<b>Mapping with Programme Outcomes</b>					
<b>COs</b>	<b>PO1</b>	<b>PO2</b>	<b>PO3</b>	<b>PO4</b>	<b>PO5</b>
<b>CO1</b>	S	S	M	M	M
<b>CO2</b>	M	M	S	M	M
<b>CO3</b>	S	S	M	S	S
<b>CO4</b>	S	M	S	M	S
<b>CO5</b>	S	M	S	M	S

\*S-Strong; M-Medium; L-Low

## PRINCIPLES OF ECONOMICS II

### Course Objectives:

To provide a frame work of knowledge relating to the concepts and practice of Economics in Indian context and to make the students understand the theories and growth of Economy in the strategic sector. Also, to provide insight on the most pressing issue “Demand for Defence Expenditure” i.e. the purpose of Defence Budget.

### Expected Course Outcomes:

On the successful completion of the course, student will be able to:

1	To learn about the basic terms in economics	K1
2	Understand about the theories and analysis in economics.	K2
3	Identify the systems used in economics	K2
4	Acquire knowledge about the strategies in economics	K3
5	To know about the budget used for Defence purpose	K4

**K1** - Remember; **K2** - Understand; **K3** - Apply; **K4** - Analyze; **K5** - Evaluate; **K6** - Create

#### Unit:1

Money - Functions - Changes in supply of Money, Inflation - Deflation - Types - Characteristics - Causes - Effects - Remedies - Deflationary Gap

#### Unit:2

Keynesian Theory of Employment - Savings and Investment Analysis

#### Unit:3

Macro-Economic Goals and Tools - Objectives - Central and Commercial Banking in India - Fiscal Policy - Indian Tax System –Recent Trends in Indian Tax Systems.

#### Unit:4

Agriculture – Role of Agriculture – Industry – Types – Classification – Role of Industries – Green Revolution – Services Sector.

<b>Unit:5</b>	
Budget - Salient Features - Central and State - Deficit Financing - Priorities in Budgeting - Expenditure in Defence - Recent Trends in Military Finance - Role of Social and Economic infrastructure for Defence Purposes.	
<b>Reference</b>	
1	Dr. S. Sankaran, Macro Economics
2	M. L. Jhingan, Macro Economic Theory
3	Dutt and Sundaram, Indian Economy
4	B. P. Tyagi, Public Economics.
5	Dr. S. Sankaran, Monetary Economics
6	Chrystal, A, Lipsey, R.G Introduction to Positive Economics, Oxford University
7	Garfinkel, M.R. (University of California) Economics of Conflict, An Overview (Paper Presentation).
8	Hartley and Sandler, Hand Book of Defence Economics, North Holland

<b>Mapping with Programme Outcomes</b>					
<b>COs</b>	<b>PO1</b>	<b>PO2</b>	<b>PO3</b>	<b>PO4</b>	<b>PO5</b>
<b>CO1</b>	S	S	M	M	M
<b>CO2</b>	M	M	S	M	M
<b>CO3</b>	S	S	M	S	S
<b>CO4</b>	S	M	S	M	S
<b>CO5</b>	S	M	S	M	S

\*S-Strong; M-Medium; L-Low

<b>COURSE TITLE</b>	<b>PHYSICS – I</b>
<b>CREDITS</b>	3
<b>COURSE OBJECTIVES</b>	To impart basic principles of Physics that which would be helpful for students who have taken programmes other than Physics.

<b>UNITS</b>	<b>COURSE DETAILS</b>
<b>UNIT-I</b>	<b>WAVES, OSCILLATIONS AND ULTRASONICS:</b> Definition of simple harmonic motion (SHM) – laws of transverse vibrations of strings – determination of AC frequency using sonometer (steel and brass wires) – ultrasound – production – piezoelectric method – application of ultrasonics: medical field – lithotripsy, ultrasonography.
<b>UNIT-II</b>	<b>PROPERTIES OF MATTER:</b> <i>Elasticity:</i> elastic constants – bending of beam – theory of non- uniform bending – determination of Young’s modulus by non-uniform bending – energy stored in a stretched wire – torsion of a wire – determination of rigidity modulus by torsional pendulum <i>Viscosity:</i> streamline and turbulent motion – critical velocity – coefficient of viscosity – Poiseuille’s formula – comparison of viscosities – burette method, <i>Surface tension:</i> definition– drop weight method – interfacial surface tension.
<b>UNIT-III</b>	<b>HEAT AND THERMODYNAMICS:</b> Joule-Kelvin effect – Joule-Thomson porous plug experiment – theory – temperature of inversion – liquefaction of Oxygen– Linde’s process of liquefaction of air– liquid Oxygen for medical purpose– importance of cryocoolers– entropy – change of entropy in reversible and irreversible process.
<b>UNIT-IV</b>	<b>ELECTRICITY AND MAGNETISM:</b> potentiometer – principle – measurement of thermo emf using potentiometer –magnetic field due to a current carrying conductor – Biot-Savart’s law – field along the axis of the coil carrying current – peak, average and RMS values of ac current and voltage .
<b>UNIT-V</b>	<b>DIGITAL ELECTRONICS AND DIGITAL INDIA:</b> logic gates, OR, AND, NOT, NAND, NOR , EXOR logic gates – universal building blocks – Boolean algebra – De Morgan’s theorem – verification – overview of Government initiatives: software technological parks under MeitY, NIELIT- semiconductor laboratories under Dept. of Space – an introduction to Digital India.
<b>PROFESSIONAL COMPONENTS</b>	Expert lectures –seminars — webinars – industry inputs – social accountability – patriotism
<b>TEXT BOOKS</b>	<ol style="list-style-type: none"> <li>1. R.Murugesan (2001), AlliedPhysics,S. ChandandCo,NewDelhi.</li> <li>2. BrijlalandN.Subramanyam (1994), WavesandOscillations,VikasPublishing House,NewDelhi.</li> <li>3. BrijlalandN.Subramaniam (1994), PropertiesofMatter,S.ChandandCo.,NewDelhi.</li> <li>4. J.B.Rajam and C.L.Arora (1976). Heat and Thermodynamics (8<sup>th</sup> edition), S.ChandandCo.,New Delhi.</li> <li>5. R.Murugesan(2005),</li> </ol>

	<p>Optics and Spectroscopy, S. Chand and Co, New Delhi.</p> <p>6. A. Subramaniam, Applied Electronics 2<sup>nd</sup> Edn., National Publishing Co., Chennai.</p>
<b>REFERENCE BOOKS</b>	<p>1. Resnick Halliday and Walker (2018). Fundamentals of Physics (11<sup>th</sup> edition), John Wiley and Sons, Asia Pvt. Ltd., Singapore.</p> <p>2. V.R. Khanna and R.S. Bedi (1998), Textbook of Sound 1<sup>st</sup> Edn. Kedharnaath Publish and Co, Meerut.</p> <p>3. N.S. Khare and S.S. Srivastava (1983), Electricity and Magnetism 10<sup>th</sup> Edn., Atma Ram and Sons, New Delhi.</p> <p>4. D.R. Khanna and H.R. Gulati (1979). Optics, S. Chand and Co. Ltd., New Delhi.</p> <p>5. V.K. Metha (2004). Principles of electronics 6<sup>th</sup> Edn. S. Chand and company.</p>
<b>WEB RESOURCES</b>	<p>1. <a href="https://youtu.be/M_5KYncYNyc">https://youtu.be/M_5KYncYNyc</a></p> <p>2. <a href="https://youtu.be/ljJLJgIvaHY">https://youtu.be/ljJLJgIvaHY</a></p> <p>3. <a href="https://youtu.be/7mGqd9HQ_AU">https://youtu.be/7mGqd9HQ_AU</a></p> <p>4. <a href="https://youtu.be/h5jOAw57OXM">https://youtu.be/h5jOAw57OXM</a></p> <p>5. <a href="https://learningtechnologyofficial.com/category/fluid-mechanics-lab/">https://learningtechnologyofficial.com/category/fluid-mechanics-lab/</a></p> <p>6. <a href="http://hyperphysics.phy-astr.gsu.edu/hbase/permot2.html">http://hyperphysics.phy-astr.gsu.edu/hbase/permot2.html</a> <a href="https://www.youtube.com/watch?v=gT8Nth9NWPM">https://www.youtube.com/watch?v=gT8Nth9NWPM</a> <a href="https://www.youtube.com/watch?v=9mXOMzUruMQ&amp;t=1s">https://www.youtube.com/watch?v=9mXOMzUruMQ&amp;t=1s</a> <a href="https://www.youtube.com/watch?v=m4u-SuaSu1sandt=3s">https://www.youtube.com/watch?v=m4u-SuaSu1sandt=3s</a> <a href="https://www.biolinscientific.com/blog/what-are-surfactants-and-how-do-they-work">https://www.biolinscientific.com/blog/what-are-surfactants-and-how-do-they-work</a></p>

### **METHOD OF EVALUATION:**

Continuous Internal Assessment	End Semester Examination	Total	Grade
25	75	100	

### **COURSE OUTCOMES:**

At the end of the course, the student will be able to:

COURSE OUTCOMES	CO1	Explain types of motion and extend their knowledge in the study of various dynamic motions analyze and demonstrate mathematically. Relate theory with practical applications in medical field.
	CO2	Explain their knowledge of understanding about materials and their behaviors and apply it to various situations in laboratory and real life. Connect droplet theory with Corona transmission.
	CO3	Comprehend basic concept of thermodynamics concept of entropy and associated theorems able to interpret the process of flow temperature physics in the background of growth of this technology.





<b>COURSE TITLE</b>	<b>PHYSICS –II</b>
<b>COURSE OBJECTIVES</b>	To understand the basic concepts of optics, modern Physics, concepts of relativity and quantum physics, semiconductor physics, and electronics.

<b>UNITS</b>	<b>COURSE DETAILS</b>
<b>UNIT-I</b>	<b>OPTICS:</b> Definition of interference – air wedge – determination of diameter of a thin wire by air wedge – diffraction – diffraction of light vs sound – normal incidence – experimental determination of wavelength using diffraction grating (no theory) – polarization – polarization by double refraction – Brewster’s law .
<b>UNIT-II</b>	<b>ATOMIC PHYSICS:</b> Mass number – atomic number – nucleons – vector atom model – various quantum numbers – Pauli’s exclusion principle – electronic configuration – periodic classification of elements –photo electric effect – Einstein’s photoelectric equation – applications of photoelectric effect: solar cells,LED, photodiode.
<b>UNIT-III</b>	<b>NUCLEAR PHYSICS:</b> Magic numbers – shell model – nuclear energy – mass defect – binding energy – radioactivity – uses – half life – mean life - radio isotopes and uses –controlled and uncontrolled chain reaction – nuclear fission – energy released in fission – critical size- atom bomb – nuclear fusion – thermonuclear reactions – differences between fission and fusion.
<b>UNIT-IV</b>	<b>INTRODUCTION TO RELATIVITY AND GRAVITATIONAL WAVES:</b> frame of reference – postulates of special theory of relativity – Galilean transformation equations – Lorentz transformation equations – derivation – length contraction – time dilation – twin paradox – mass-energy equivalence.
<b>UNIT-V</b>	<b>SEMICONDUCTOR PHYSICS:</b> p-n junction diode – forward and reverse biasing – characteristic of diode – zener diode – characteristic of zener diode – voltage regulator – full wave bridge rectifier – construction and working – advantages (no mathematical treatment) – USB cell phone charger –introduction to e-vehicles and EV charging stations.
<b>TEXT BOOKS</b>	<ol style="list-style-type: none"> <li>1. R.Murugesan (2005), AlliedPhysics,S.ChandandCo,NewDelhi.</li> <li>2. K.ThangarajandD.Jayaraman(2004), AlliedPhysics,PopularBookDepot,Chennai.</li> <li>3. BrijlalandN.Subramanyam(2002), TextbookofOptics,S.ChandandCo,NewDelhi.</li> <li>4. R.Murugesan (2005), ModernPhysics,S.ChandandCo,NewDelhi.</li> <li>5. A.SubramaniamAppliedElectronics, 2<sup>nd</sup>Edn.,NationalPublishingCo.,Chennai.</li> </ol>
<b>REFERENCE BOOKS</b>	<ol style="list-style-type: none"> <li>1. ResnickHallidayandWalker (2018), FundamentalsofPhysics, 11<sup>th</sup>Edn.,JohnWileyandSons, Asia Pvt.Ltd.,Singapore.</li> <li>2. D.R.KhannaandH.R. Gulati (1979).Optics, S.ChandandCo.Ltd.,New Delhi.</li> <li>3. A.Beiser (1997), ConceptsofModernPhysics,TataMcGrawHillPublication,NewDelhi.</li> </ol>

	<p>4. Thomas L. Floyd (2017), Digital Fundamentals, 11<sup>th</sup>Edn., Universal Book Stall, NewDelhi.</p> <p>5. V.K.Metha(2004), Principlesofelectronics, 6<sup>th</sup>Edn., S.Chandand Company, New Delhi.</p>
<b>WEB RESOURCES</b>	<p>1. <a href="https://www.berkshire.com/learning-center/delta-p-facemask/https://www.youtube.com/watch?v=OrhxU47gtj4https://www.youtube.com/watch?time_continue=31&amp;v=D38BjgUdL5U&amp;feature=emb_logo">https://www.berkshire.com/learning-center/delta-p-facemask/https://www.youtube.com/watch?v=OrhxU47gtj4https://www.youtube.com/watch?time_continue=31&amp;v=D38BjgUdL5U&amp;feature=emb_logo</a></p> <p>2. <a href="https://www.youtube.com/watch?v=JrRrp5F-Qu4">https://www.youtube.com/watch?v=JrRrp5F-Qu4</a></p> <p>3. <a href="https://www.validyne.com/blog/leak-test-using-pressure-transducers/">https://www.validyne.com/blog/leak-test-using-pressure-transducers/</a></p> <p>4. <a href="https://www.atoptics.co.uk/atoptics/blsky.htm">https://www.atoptics.co.uk/atoptics/blsky.htm</a> -</p> <p>5. <a href="https://www.metoffice.gov.uk/weather/learn-about/weather/optical-effects">https://www.metoffice.gov.uk/weather/learn-about/weather/optical-effects</a></p>

METHOD OF EVALUATION:

Continuous InternalAssessment	End Semester Examination	Total	Grade
25	75	100	

COURSE OUTCOMES:

Attheendofthecourse,the studentwillbeableto:

<b>COURSE OUTCOMES</b>	<b>CO1</b>	Explaintheconceptsof interferencediffractionusingprinciplesof superpositionofwaves and rephrase the concept of polarization based on wave patterns
	<b>CO2</b>	Outline the basic foundation of different atom models and various experiments establishing quantum concepts. Relate the importance of interpreting improving theoretical models based on observation. Appreciate interdisciplinary nature of science and in solar energy related applications.
	<b>CO3</b>	Summarizethepropertiesofnuclei, nuclearforcesstructureofatomicnucleusandnuclear models. Solveproblems on delayratehalf-lifeand mean-life. Interpret nuclear processes like fission and fusion. Understand the importance of nuclear energy, safety measures carried and get our Govt.agencies like DAE guiding the country in the nuclear field.
	<b>CO4</b>	Todescribethebasicconceptsofrelativitylikeequivalenceprinciple, inertialframes and Lorentz transformation. Extend their knowledge on concepts ofrelativityandvice-versa. Relate this with current research in this field and get an overview of research projects of National and International importance, like LIGO, ICTS, and opportunities available.
	<b>CO5</b>	Summarize the working of semiconductor devices like junction diode, Zener diode, transistors and practical devices we daily use like USB chargers and EV charging stations.



<b>COURSE</b>	<b>CORE</b>
<b>COURSE TITLE</b>	<b>PHYSICS PRACTICALS</b>
<b>COURSE OBJECTIVES</b>	Apply various physics concepts to understand properties of matter, light, electricity, and electronics by experimentation to verify theories, quantify and analyses- impart the skill to do error analysis and correlate the results
<p><b>Minimum of 8 Experiments from the list:</b></p> <ol style="list-style-type: none"> <li>1. Young's modulus by non-uniform bending using pin and microscope.</li> <li>2. Rigidity modulus by torsional oscillations without mass.</li> <li>3. Surface tension and interfacial Surface tension – drop weight method.</li> <li>4. Comparison of viscosities of two liquids – burette method.</li> <li>5. Determination of g by compound pendulum.</li> <li>6. Thickness of a wire using air wedge.</li> <li>7. Refractive index of liquid using liquid prism (hollow prism).</li> <li>8. Determination of AC frequency using sonometer.</li> <li>9. Calibration of low range voltmeter using potentiometer.</li> <li>10. Specific resistance of a wire using PO box.</li> <li>11. Determination of figure of merit of table galvanometer.</li> <li>12. Verification of truth tables of basic logic gates using ICs.</li> <li>13. Characterisation of Zener diode (Forward and Reverse Bias).</li> <li>14. NAND as universal building block (AND, OR, NOT gates).</li> <li>15. NOR gate as a universal building block (AND, OR, NOT gates).</li> </ol> <p><i>Note</i> : Use of digital balance permitted</p>	

METHOD OF EVALUATION:

<b>Continuous Internal Assessment</b>	<b>End Semester Examination</b>	<b>Total</b>	<b>Grade</b>
25	75	100	

## CHEMISTRY - I

<b>Prerequisites</b>	Higher secondary chemistry
<b>Objectives of the course</b>	<p>This course aims to provide knowledge on the</p> <ul style="list-style-type: none"> <li>• basics of atomic orbitals, chemical bonds, hybridization</li> <li>• concepts of thermodynamics and its applications.</li> <li>• concepts of nuclear chemistry</li> <li>• importance of chemical industries</li> <li>• Qualitative and analytical methods.</li> </ul>
<b>Course Outline</b>	<p><b>UNIT I</b>  <b>Chemical Bonding and Nuclear Chemistry</b></p> <p>Chemical Bonding: Molecular Orbital Theory-bonding, antibonding and non-bonding orbitals. Molecular orbital diagrams for Hydrogen, Helium, Nitrogen; discussion of bond order and magnetic properties.</p> <p>Nuclear Chemistry: Fundamental particles - Isotopes, Isobars, Isotones and Isomers-Differences between chemical reactions and nuclear reactions - group displacement law. Nuclear binding energy - mass defect - calculations. Nuclear fission and nuclear fusion - differences – Stellar energy. Applications of radioisotopes - carbon dating, rock dating and medicinal applications.</p>

	<p><b>Unit II</b>  <b>Industrial Chemistry</b></p> <p>Fuels: Fuel gases: Natural gas, water gas, semi water gas, carbureted water gas, producer gas, CNG, LPG and oil gas (manufacturing details not required). Silicones: Synthesis, properties and uses of silicones.</p> <p>Fertilizers: Urea, ammonium sulphate, potassium nitrate, NPK fertilizer, superphosphate, triple superphosphate.</p>
	<p><b>UNIT III</b>  <b>Fundamental Concepts in Organic Chemistry</b></p> <p>Hybridization: Orbital overlap, hybridization and geometry of CH<sub>4</sub>, C<sub>2</sub>H<sub>4</sub>, C<sub>2</sub>H<sub>2</sub> and C<sub>6</sub>H<sub>6</sub>. Electronic effects: Inductive effect and consequences on K<sub>a</sub> and K<sub>b</sub> of organic acids and bases, electromeric, mesomeric, hyper conjugation and steric- examples.</p> <p>Reaction mechanisms: Types of reactions–aromaticity (Huckel’s rule) – aromatic electrophilic substitution; nitration, halogenation, Friedel-Craft’s alkylation and acylation. Heterocyclic compounds: Preparation, properties of pyrrole and pyridine.</p>

	<p><b>UNIT IV</b></p> <p><b>Thermodynamics and Phase Equilibria</b></p> <p>Thermodynamics: Types of systems, reversible and irreversible processes, isothermal and adiabatic processes and spontaneous processes. Statements of first law and second law of thermodynamics. Carnot's cycle and efficiency of heat engine. Entropy and its significance. Free energy change and its importance (no derivation).</p> <p>Conditions for spontaneity in terms of entropy and Gibbs free energy.</p> <p>Relationship between Gibbs free energy and entropy.</p> <p>Phase Equilibria: Phase rule - definition of terms in it. Applications of phase rule to water system. Two component system - Reduced phase rule and its application to a simple eutectic system (Pb-Ag).</p>
	<p><b>UNIT V</b></p> <p><b>Analytical Chemistry</b></p> <p>Introduction to qualitative and quantitative analysis. Principles of volumetric analysis. Separation and purification techniques – extraction, distillation and crystallization.</p> <p>Chromatography: principle and application of column, paper and thin layer chromatography.</p>
<p>Extended Professional Component (is a part of internal component only, Not to be included in the external examination question paper)</p>	<p>Questions related to the above topics, from various competitive examinations UPSC/ JAM /TNPSC others to be solved (To be discussed during the Tutorial hours)</p>
<p>Skills acquired from this course</p>	<p>Knowledge, Problem solving, Analytical ability, Professional Competency, Professional Communication and Transferable skills.</p>
<p><b>Recommended Text</b></p>	<ol style="list-style-type: none"> <li>1. V.Veeraiyan, Text book of Ancillary Chemistry; High mount publishing house, Chennai, first edition,2009.</li> <li>2. S.Vaithyanathan, Text book of Ancillary Chemistry; Priya Publications, Karur,2006.</li> <li>3. S.ArunBahl, B.S.Bahl, Advanced Organic Chemistry; S.Chand and Company, NewDelhi, twenty third edition, 2012.</li> <li>4. P.L.Soni, H.M.Chawla, Text Book of Organic Chemistry; Sultan Chand &amp; sons, New Delhi, twenty ninth edition, 2007.</li> </ol>

<b>Reference Books</b>	<p>5. P.L.Soni, Mohan Katyal, Textbook of Inorganic chemistry; Sultan Chand Company, New Delhi, twentieth edition, 2007.</p> <p>6. B.R.Puri, L.R.Sharma, M.S.Pathania, Textbook Physical Chemistry; Vishal Publishing Co., New Delhi, fortyfourth edition, 2018.</p> <p>7. B.K, Sharma, Industrial Chemistry; GOEL publishing house, Meerut, sixteenth edition, 2014.</p>
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**Course Learning Outcomes (for Mapping with POs and PSOs) On completion of the course the students should be able to**

CO 1: gain in-depth knowledge about the theories of chemical bonding, nuclear reactions and its applications.

CO 2: evaluate the efficiencies and uses of various fuels and fertilizers

CO 3: explain the type of hybridization, electronic effect and mechanism involved in the organic reactions.

CO 4: apply various thermodynamic principles, systems and phase rule.

CO 5: explain various methods to identify an appropriate method for the separation of chemical components

CO /PSO	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	3	3	3	3	3
CO2	3	3	3	3	3
CO3	3	3	3	3	3
CO4	3	3	3	3	3
CO5	3	3	3	3	3
<b>Weightage</b>	15	15	15	15	15
<b>Weighted percentage of Course Contribution to POs</b>	3.0	3.0	3.0	3.0	3.0

**Level of Correlation between PSO's and CO's**

CO /PO	PO1	PO2	PO3	PO4	PO5
CO1	3	3	3	3	3
CO2	3	3	3	3	3
CO3	3	3	3	3	3
CO4	3	3	3	3	3
CO5	3	3	3	3	3
<b>Weightage</b>	15	15	15	15	15
<b>Weighted percentage of Course Contribution to POs</b>	3.0	3.0	3.0	3.0	3.0

**Level of Correlation between PO's and CO's**

<b>Title of the Course</b>	<b>CHEMISTRY II</b>
<b>Prerequisites</b>	Chemistry for physical sciences -I
<b>Objectives of the course</b>	<p>This course aims at providing knowledge on the</p> <ul style="list-style-type: none"> <li>• Co-ordination Chemistry and Water Technology</li> <li>• Carbohydrates and Amino acids</li> <li>• basics and applications of electrochemistry</li> <li>• basics and applications of kinetics and catalysis</li> <li>• Various photochemical phenomenon</li> </ul>
<b>Course Outline</b>	<p><b>UNIT I</b>  <b>Co-ordination Chemistry and Water Technology</b>  Co-ordination Chemistry: Definition of terms-IUPAC  Nomenclature - Werner's theory - EAN rule - Pauling's theory - Postulates - Applications to <math>[\text{Ni}(\text{CO})_4]</math>, <math>[\text{Ni}(\text{CN})_4]^{2-}</math>, <math>[\text{Co}(\text{CN})_6]^{3-}</math> Chelation - Biological role of Haemoglobin and Chlorophyll (elementary idea) - Applications in qualitative and quantitative analysis.</p> <p>Water Technology: Hardness of water, determination of hardness of water using EDTA method, zeolite method- Purification techniques- BOD, COD.</p> <p><b>Unit II</b>  <b>Carbohydrates and Amino acids</b>  Carbohydrates: Classification, preparation and properties of glucose, fructose and sucrose. Discussion of open chain ring structures of glucose and fructose. Glucose -fructose interconversion. Properties of starch and cellulose.</p> <p>Amino acids: Classification - preparation and properties of alanine, preparation of dipeptides using Bergmann method. RNA and DNA (elementary idea only).</p> <p><b>UNIT III</b>  <b>Electrochemistry</b>  Galvanic cells - Standard hydrogen electrode - calomel electrode - standard electrode potentials -electrochemical series. Strong and weak electrolytes - ionic product of water -pH, pKa, pKb. Conductometric titrations - pH determination by colorimetric method - buffer solutions and its biological applications - electroplating - Nickel and chrome plating - Types of cells -fuel cells-corrosion and its prevention.</p>



	<p><b>UNIT IV</b> <b>Kinetics and Catalysis</b></p> <p>Order and molecularity. Integrated rate expression for I and II (2A <math>\square</math> Products) order reactions. Pseudo first order reaction, methods of determining order of a reaction – Half-life period – Catalysis - homogeneous and heterogeneous, catalyst used in Contact and Haber’s processes. Concept of energy of activation and Arrhenius equation.</p> <p><b>UNIT V</b> <b>Photochemistry</b></p> <p>Grothus-Draper’s law and Stark-Einstein’s law of photochemical equivalence, Quantum yield - Hydrogen-chloride reaction. Phosphorescence, fluorescence, chemiluminescence and photosensitization and photosynthesis (definition with examples).</p>
<p>Extended Professional Component (is a part of internal component only, Not to be included in the external examination question paper)</p>	<p>Questions related to the above topics, from various competitive examinations UPSC/ JAM /TNPSC others to be solved (To be discussed during the Tutorial hours)</p>
<p>Skills acquired from this course</p>	<p>Knowledge, Problem solving, Analytical ability, Professional Competency, Professional Communication and Transferable skills.</p>
<p><b>Recommended Text</b></p>	<ol style="list-style-type: none"> <li>1. V.Veeraiyan, Textbook of Ancillary Chemistry; High mount publishing house, Chennai, first edition,2009.</li> <li>2. S.Vaithyanathan, Text book of Ancillary Chemistry; Priya Publications, Karur,2006.</li> <li>3. Arun Bahl, B.S.Bahl, Advanced Organic Chemistry; S.Chand and Company, New Delhi, twenty third edition, 2012.</li> <li>4. P.L.Soni, H.M.Chawla, Text Book of Organic Chemistry; Sultan Chand &amp; sons, New Delhi, twenty ninth edition, 2007.</li> </ol>
<p><b>Reference Books</b></p>	<ol style="list-style-type: none"> <li>1. P.L.Soni, Mohan Katyal, Text book of Inorganic chemistry; Sultan Chand and Company, New Delhi, twentieth edition, 2007.</li> <li>2. R.Puri, L.R.Sharma, M.S.Pathania, Text book Physical Chemistry; Vishal Publishing Co., New Delhi, forty seventh edition, 2018.</li> </ol>

	3. B.K,Sharma, Industrial Chemistry; GOEL publishing house, Meerut, sixteenth edition, 2014.
<b>Website and e-learning source</b>	
<b>Course Learning Outcomes (for Mapping with POs and PSOs)</b> <b>On completion of the course the students should be able to</b> <b>CO 1:</b> write the IUPAC name for complex, different theories to explain the bonding in coordination compounds and water technology <b>CO 2:</b> explain the preparation and property of carbohydrate, amino acids and nucleic acids. <b>CO 3:</b> apply/demonstrate the electrochemistry principles in corrosion, electroplating and fuel cells. <b>CO 4:</b> identify the reaction rate, order for chemical reaction and explain the purpose of a catalyst. <b>CO 5:</b> outline the various type of photochemical process.	

CO /PSO	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	3	3	3	3	3
CO2	3	3	3	3	3
CO3	3	3	3	3	3
CO4	3	3	3	3	3
CO5	3	3	3	3	3
<b>Weightage</b>	15	15	15	15	15
<b>Weighted percentage of Course Contribution to PSOs</b>	3.0	3.0	3.0	3.0	3.0

#### Level of Correlation between PSO's and CO's

CO /PO	PO1	PO2	PO3	PO4	PO5
CO1	3	3	3	3	3
CO2	3	3	3	3	3
CO3	3	3	3	3	3
CO4	3	3	3	3	3
CO5	3	3	3	3	3
<b>Weightage</b>	15	15	15	15	15
<b>Weighted percentage of Course Contribution to POs</b>	3.0	3.0	3.0	3.0	3.0

#### Level of Correlation between PO's and CO's

<b>Title of the Course</b>	<b>CHEMISTRY II Practical</b>
<b>Objectives of the course</b>	<p>This course aims to provide knowledge on</p> <ul style="list-style-type: none"> <li>• identification of organic functional groups</li> <li>• different types of organic compounds with respect to their properties.</li> <li>• determination of elements in organic compounds..</li> </ul>
	<b>SYSTEMATIC ANALYSIS OF ORGANIC COMPOUNDS</b> The analysis must be carried out as follows:
	<p>(a) Functional group tests [phenol, acids (mono &amp; di) aromatic primary amine, amides (mono &amp; di), aldehyde and glucose].</p> <p>(b) Detection of elements (N, S, Halogens).</p> <p>(c) To distinguish between aliphatic and aromatic compounds.</p> <p>(d) To distinguish – Saturated and unsaturated compounds.</p>
<b>Reference Books</b>	V.Venkateswaran, R.Veerasingam, A.R.Kulandaivelu, Basic Principles of Practical Chemistry; Sultan Chand & sons, Second edition, 1997.

**Course Learning Outcomes (for Mapping with POs and PSOs)**

**On completion of the course the students should be able to**

CO 1: gain an understanding of the use of standard flask and volumetric pipettes, burette.

CO 2: design, carry out, record and interpret the results of volumetric titration.

CO 3: apply their skill in the analysis of water/hardness.

CO4: analyze the chemical constituents in allied chemical products

CO /PSO	PSO1	PSO2	PSO3	PSO4	PSO5
CO1	3	3	3	3	3
CO2	3	3	3	3	3
CO3	3	3	3	3	3
CO4	3	3	3	3	3
<b>Weightage</b>	12	12	12	12	12
<b>Weighted percentage of Course Contribution to PSOs</b>	3.0	3.0	3.0	3.0	3.0

**Level of Correlation between PSO's and CO's**

CO /PO	PO1	PO2	PO3	PO4	PO5
CO1	3	3	3	3	3
CO2	3	3	3	3	3
CO3	3	3	3	3	3
CO4	3	3	3	3	3
<b>Weightage</b>	12	12	12	12	12
<b>Weighted percentage of Course Contribution to POs</b>	3.0	3.0	3.0	3.0	3.0

**Level of Correlation between PO's and CO's**

Title of the Course	DISCRETE MATHEMATICS-I	
<b>Learning Objective</b>		
LO1	To make the students understand the Mathematical Logic and truth table.	
LO2	To know about how and when to use set theory.	
LO3	To understand the discrete structure, storage structure.	
LO4	To understand the methods of Relations and ordering.	
LO5	To understand the functions, classifications, and types.	
UNIT	Contents	No. of Hours
I	<b>Mathematical logic-:</b> Connectives, well formed formulas, Tautology, Equivalence of formulas, Tautological implications, Duality law, Normal forms.	12
II	<b>Set Theory:</b> Basic Concept of Set Theory – Operations on Sets – Venn Diagram	12
III	Representation of Discrete Structure : Data Structure – Storage Structure - Sequential Allocation – Pointers and Linked Allocation – An Application of Bit Represented Sets.	12
IV	<b>Relations and Ordering:</b> Relations – Properties of Binary Relations in a set – Relation Matrix and the Graph of a Relation – Partition and Covering of a set – Equivalence Relations – Compatibility Relations – Composition of Binary Relations – Partial Ordering – Partially Ordered set.	12
V	<b>Functions</b> Definitions of functions and its Classification – Types – Examples – Composition of functions – Inverse functions – Binary and n-ary operations – Characteristic function of a set – Hashing functions – Recursive functions	12
<b>Total</b>		60
Course Outcomes		Programme Outcome
CO	Solve problems in Mathematical logic and truth table.	
1	Know and understand about set theory.	PO1, PO6
2	Know and understand about discrete structure, storage structure.	PO2
3	Know and understand about Relations and Ordering	PO4, PO5
4	Understand the functions, classification and types.	PO6
<b>Text Book</b>		
Discrete Mathematical Structures with applications to computer Science J.P Tremblay and R.P Manohar (Mc.Graw Hill, 1997.)		
<b>Reference Books</b>		
P.R. Vittal, Mathematical Foundations– Margham Publication, Chennai. Discrete Mathematics-Oscar Levin(3rd Edition)		
<b>Web Resources</b>		
<a href="https://nptel.ac.in/courses/106106094">https://nptel.ac.in/courses/106106094</a>		
<a href="https://nptel.ac.in/courses/111107058">https://nptel.ac.in/courses/111107058</a>		

Mapping with Programme Outcomes:

<b>CO/PSO</b>	<b>PSO 1</b>	<b>PSO 2</b>	<b>PSO 3</b>	<b>PSO 4</b>	<b>PSO 5</b>	<b>PSO 6</b>
<b>CO 1</b>	3	3	3	3	3	3
<b>CO 2</b>	3	2	3	3	2	3
<b>CO 3</b>	3	3	3	3	3	3
<b>CO 4</b>	3	2	2	3	3	3
<b>CO 5</b>	3	2	3	3	3	2
<b>Weight age of course contributed to each PSO</b>	15	12	14	15	14	14

S-Strong-3 M-Medium-2 L-Low-1

<b>Title of the Course</b>	<b>DISCRETE MATHEMATICS - II</b>
Objectives of the Course	This course aims to develop mathematical maturity and ability to deal with abstraction and to develop construction and verification of formallogical manipulation.
Course Outline	<p>UNIT I: RECURRENCE RELATIONS AND GENERATING FUNCTIONS  Recurrence - Polynomials and their Evaluations - Recurrence Relations - Solution of Finite Order Homogeneous [linear] Relations - Solutions of Non-homogeneous Relations.  (Chap V . Sections:1 to 5 )</p> <p>UNIT II: MATHEMATICAL LOGIC  TF Statements - Connectives - Atomic and Compound Statements - Well-formed [StatementFormulae]- Truth Table of a Formula-Tautology -Tautological Implications and Equivalence of Formulae.  (Chap IX . Sections:1 to 8 )</p> <p>UNIT III: MATHEMATICAL LOGIC [CONTD..]  Replacement process - Functionally complete sets of connectives and Duality law – NormalForms-PrincipalNormalForms.(Chap IX . Sections: 9 to 12 )</p> <p>UNIT IV: LATTICES  Lattices [omit example 15 PpNo.10.6]- Some properties of Lattices - New Lattices (omit remark Pp 10.14)-Modular and Distributive Lattices (omit theorem 10 and 17,Example 4-Pp10.23, Example 11-Pp10.24)  (Chap X . Sections:1 to 4 )</p> <p>UNIT-V BOOLEAN ALGEBRA  Boolean Algebra (omit theorem 25) - Boolean Polynomials- Karnaugh Maps (omit K- map for 5 and 6 variables) (Chap X . Sections:5 to 7)</p>
Skills acquired from this course	Knowledge, Problem Solving, Analytical ability, Professional Competency, Professional Communication and Transferrable Skill
Recommended Text	M.K.Venkataraman, N.Sridharan and N.Chandrasekaran, [2003] Discrete Mathematics, The National Publishing company, chennai.
Reference Books	<p>Oscar Levin, Discrete Mathematics, 3rd Edition,2016.</p> <p>B. A. Davey &amp; H. A. Priestley (2002). <i>Introduction to Lattices and Order</i> (2<sup>nd</sup> edition). Cambridge University Press.</p> <p>Edgar G. Goodaire&amp; Michael M. Parmenter (2018). <i>Discrete Mathematics with Graph Theory</i> (3rd edition). Pearson Education.</p> <p>Rudolf Lidl&amp; Günter Pilz (1998). <i>Applied Abstract Algebra</i> (2nd edition). Springer.</p> <p>Kenneth H. Rosen (2012). <i>Discrete Mathematics and its Applications: With Combinatorics and Graph Theory</i> (7th edition). McGraw-Hill.</p> <p>C. L. Liu (1985). <i>Elements of Discrete Mathematics</i> (2nd edition). McGraw-Hill.</p>
Website and e-Learning Source	<a href="https://nptel.ac.in">https://nptel.ac.in</a>

**Course Learning Outcomes:**

This course will enable the students to:

CO Number	CO Statement	Knowledge Level
CO1	Analyse and perceive various graph theoretic concepts and familiarize with their applications.	K4, K5
CO2	Describe about partially ordered sets, Boolean algebra, lattices and their types.	K1
CO3	Apply Karnaugh map for simplifying the Boolean expression	K3
CO4	Demonstrate the skill to construct simple mathematical proofs and to validate.	K2, K6
CO5	Achieve greater accuracy, clarity of thought and language.	K6

CO	Programme Outcomes (PO)							Programme Specific Outcomes (PSO)					Mean Scores of COs
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PSO1	PSO2	PSO3	PSO4	PSO5	
1	2	3	2	3	3	3	3	3	3	3	2	2	2.67
2	3	2	2	2	3	3	3	2	2	2	2	3	2.42
3	2	2	2	2	3	3	3	3	3	3	2	3	2.58
4	3	2	2	3	3	3	2	3	3	3	3	2	2.67
5	3	2	2	3	3	2	2	3	3	2	2	3	2.5
	Mean Overall Score											2.57	
	Result											High	





Subject Code	Subject Name	Category	L	T	P	S	Inst. hours	Credits	Marks		
									CIA	External	Total
	<b>Statistical Methods and their Applications-I</b>	Elective	2	-	-	-	4	3	25	75	100
<b>Learning Objectives</b>											
<b>LO1</b>	Understand basic concepts of Statistical Methods										
<b>LO2</b>	Have a basic understanding of measures of location										
<b>LO3</b>	Have a basic understanding of measures of dispersion										
<b>LO4</b>	Understand about Measures of skewness										
<b>LO5</b>	Understand about correlation										
<b>UNIT</b>	<b>Contents</b>									<b>No. Of. Hours</b>	
I	Introduction - scope and limitations of statistical methods - classification of data -Tabulation of data- Diagrammatic and Graphical representation of data - Graphical determination of Quartiles ,Deciles and Percentiles.									<b>6</b>	
II	Measures of location: Arithmetic mean, median, mode, geometric mean and Harmonic mean and their properties.									<b>6</b>	
III	Measures of dispersion: Range, Quartile deviation, mean deviation, Standard deviation, combined Standard deviation, and their relative measures.									<b>6</b>	
IV	Measures of Skewness: Karl Pearson's, Bowley's, and Kelly's and co-efficient of Skewness and kurtosis based on moments.									<b>6</b>	
V	Correlation - Karl Pearson - Spearman's Rank correlation - concurrent deviation methods. Regression Analysis: Simple Regression Equations.									<b>6</b>	
									<b>TOTAL HOURS</b>		<b>30</b>
<b>Course Outcomes</b>									<b>Programme Outcomes</b>		
CO	On completion of this course, students will										
CCO1	Learn the basics of statistical methods									PO1, PO2, PO3, PO4, PO5, PO6	
CCO2	Understanding of measures of location									PO1, PO2, PO3, PO4, PO5, PO6	
CO3	understanding of measures of dispersion									PO1, PO2, PO3, PO4, PO5, PO6	

CCO4	Understand about Measures of skewness	PO1, PO2, PO3, PO4, PO5, PO6
CO5	Understand about correlation	PO1, PO2, PO3, PO4, PO5, PO6
<b>Textbooks</b>		
1		Fundamental of Mathematical Statistics-S.C.Gupta &V.K.Kapoor-SultanChand
2		Statistical Methods-Snedecor G.W.& Cochran W.G.oxford &+DII
<b>Reference Books</b>		
1.		Elements of Statistics -Mode. E.B.-Prentice Hall
2.		Statistical Methods-Dr.S.P.Gupta-Sultan Chand &Sons
<b>Web Resources</b>		
1.		<a href="https://www.simplilearn.com/what-is-statistical-analysis-article">https://www.simplilearn.com/what-is-statistical-analysis-article</a>

**Mapping with Programme Outcomes:**

CO/PSO	PSO 1	PSO 2	PSO 3	PSO 4	PSO 5	PSO 6
CO1	3	2	1	2	1	2
CO2	3	3	2	2	3	3
CO3	3	3	2	3	3	2
CO4	3	2	3	2	2	3
CO5	3	2	2	2	3	3
<b>Weightage of course contributed to each PSO</b>	<b>15</b>	<b>12</b>	<b>10</b>	<b>11</b>	<b>12</b>	<b>13</b>

S-Strong-3    M-Medium-2    L-Low-1

Subject Code	Subject Name	Category	L	T	P	S	Inst.	Cr.	Marks			
									C	T	Total	
	<b>Statistical Methods and their Applications-II</b>	Elective	2	-	-	-	3	3	25	75	100	
<b>Learning Objectives</b>												
LO1	Understand basic concepts of curve fitting.											
LO2	Have a basic understanding of Sample Space											
LO3	Have a basic understanding of standard distribution											
LO4	Understand about Test of Significance											
LO5	Understand about Analysis of variance											
<b>UNIT</b>	<b>Contents</b>										<b>No. Of. Hours</b>	
I	Curve fitting by the methods of least squares- $Y = ax + b, Y = ax^2 + bx + c, Y = axb, Y = a e^{bx}$ and $Y = abx$										<b>6</b>	
II	Sample Space-events-probability-Addition and Multiplication Theorem-conditional probability - Baye's Theorem. Mathematical expectation Addition and Multiplication theorem, Chebychev's Inequality.										<b>6</b>	
III	Standard distributions-Binomial, Poisson, Normal distribution and fitting of these distributions.										<b>6</b>	
IV	Test of Significance-small sample and large sample test based on mean, S.D. correlation and proportion- confidence interval.										<b>6</b>	
V	Analysis of variance-One and Two way classifications-Basic principle of design of Experiments- Randomisation, Replication and Local control-C.R.D., R.B.D. and L.S.D										<b>6</b>	
<b>TOTAL HOURS</b>										<b>30</b>		
<b>Course Outcomes</b>										<b>Programme Outcomes</b>		
CO	On completion of this course, students will											
CCO1	Learn the basics of curve fitting methods.										PO1, PO2, PO3, PO4, PO5, PO6	
CCO2	Understanding of Sample Space										PO1, PO2, PO3, PO4, PO5, PO6	
CO3	Understanding of standard distribution										PO1, PO2, PO3, PO4, PO5, PO6	
CCO4	Understand about Test of Significance										PO1, PO2, PO3, PO4, PO5, PO6	
CO5	Understand about Analysis of variance										PO1, PO2, PO3, PO4, PO5, PO6	
<b>Textbooks</b>												
1	Fundamental of Mathematical Statistics-S.C.Gupta & V.K.Kapoor-Sultan Chand											
2	Statistical Methods-Snedecor G.W.& Cochran W.G.oxford &+DII											
<b>Reference Books</b>												
1.	Elements of Statistics -Mode.E.B.-Prentice Hall											
2.	Statistical Methods-Dr.S.P.Gupta-Sultan Chand & Sons											

**Mapping with Programme Outcomes:**

<b>CO/PSO</b>	<b>PSO 1</b>	<b>PSO 2</b>	<b>PSO 3</b>	<b>PSO 4</b>	<b>PSO 5</b>	<b>PSO 6</b>
<b>CO1</b>	<b>3</b>	<b>2</b>	<b>1</b>	<b>2</b>	<b>1</b>	<b>2</b>
<b>CO2</b>	<b>3</b>	<b>3</b>	<b>2</b>	<b>2</b>	<b>3</b>	<b>3</b>
<b>CO3</b>	<b>3</b>	<b>3</b>	<b>2</b>	<b>3</b>	<b>3</b>	<b>2</b>
<b>CO4</b>	<b>3</b>	<b>2</b>	<b>3</b>	<b>2</b>	<b>2</b>	<b>3</b>
<b>CO5</b>	<b>3</b>	<b>2</b>	<b>2</b>	<b>2</b>	<b>3</b>	<b>3</b>
<b>Weightage of course contributed to each PSO</b>	<b>15</b>	<b>12</b>	<b>10</b>	<b>11</b>	<b>12</b>	<b>13</b>

**S-Strong-3    M-Medium-2    L-Low-1**

<b>MICROBIOLOGY I</b>			
<b>Course Objectives</b>			
CO1	Learn the History and Evolution of Microbiology.		
CO2	Describe the structural organization, morphology and reproduction of microbes.		
CO3	Explain the methods of cultivation of microbes.		
CO4	Understand the microscopy and staining techniques		
CO5	Compare and contrast the different methods of sterilization.		
<b>UNIT</b>	<b>Details</b>	<b>No.of Hours</b>	<b>Course Objectives</b>
I	<u>History and Evolution of Microbiology, Classification – Three kingdom, five kingdom and eight kingdom. Spontaneous generation – Biogenesis Contributions of Leeuwenhoek, Louis Pasteur, Robert Koch, Elie Metchnikoff and Fleming.</u>	8	CO1
II	<u>General characteristics of microorganisms -Bacteria, Algae, Fungi, Viruses and Protozoa. Differences between prokaryotic and eukaryotic microorganisms. Anatomy of prokaryotes - cell wall, cytoplasmic membrane, cilia, flagella capsule, cytoplasmic inclusions, sporulation.</u>	8	CO2
III	Bacterial culture media and pure culture techniques. Anaerobic culture techniques.	8	CO3
IV	Microscopy – Simple, bright field, dark field, phase contrast, fluorescent, electron microscope – TEM & SEM. Staining methods.	8	CO4
V	<u>Sterilization - methods of sterilization and Disinfection. Antimicrobial chemotherapy - tests for sensitivity to antimicrobial agents.</u>	8	CO5
	Total	40	

<b>Course Outcomes</b>		
<b>Course Outcomes</b>	On completion of this course, students will;	
CO1	Study the historical events that led to the discoveries and inventions and understand the Classification of Microorganisms.	PO5, PO6, PO10
CO2	Gain Knowledge of detailed structure and functions of prokaryotic cell organelles.	PO10
CO3	Understand the various microbiological techniques, different types of media, and techniques involved in culturing microorganisms.	PO11
CO4	Explain the principles and working mechanism of different microscopes/Microscope, their function and scope of application.	PO4, PO11
CO5	Understand the concept of asepsis and modes of sterilization and disinfectants.	PO4, PO11
<b>Text Books</b>		
1	Pelczar. M. J., Chan E.C.S. and Noel. R.K. (2007). Microbiology. 7 <sup>th</sup> Edition., McGraw – Hill, New York.	
2	Willey J., Sherwood L., and Woolverton C. J., (2017). Prescott’s Microbiology. 10 <sup>th</sup> Edition., McGraw-Hill International edition.	
3	Salle. A.J (1992). Fundamental Principles of Bacteriology. 7 <sup>th</sup> Edition., McGraw Hill Inc. New York.	
4	Boyd, R.F. (1998). General Microbiology, 2 <sup>nd</sup> Edition., Times Mirror, Mosby College Publishing, St Louis.	
<b>References Books</b>		
1	Jeffrey C. Pommerville., Alcamo’s Fundamentals of Microbiology (9 <sup>th</sup> Edition). Jones & Bartlett learning 2010.	
2	Stanier R.Y, Ingraham J. L., Wheelis M. L., and Painter R. R. (2010). General Microbiology, 5 <sup>th</sup> Edition., MacMillan Press Ltd	
3	Tortora, G.J., Funke, B.R. and, Case, C.L (2013). Microbiology-An Introduction, 11 <sup>th</sup> Edition., Benjamin Cummings.	
4	Nester E., Anderson D., Roberts C. E., and Nester M. (2006). Microbiology-A Human Perspective, 5 <sup>th</sup> Edition., McGraw Hill Publications.	
5	Madigan M.T., Martinko J.M., Stahl D.A, and Clark D. P. (2010). Brock - Biology of Microorganisms, 13 <sup>th</sup> Edition Benjamin-Cummings Pub Co.	
<b>Web Resources</b>		
1	<a href="https://www.cliffsnotes.com/study-guides/biology/microbiology/introduction-to-microbiology/a-brief-history-of-microbiology">https://www.cliffsnotes.com/study-guides/biology/microbiology/introduction-to-microbiology/a-brief-history-of-microbiology</a>	
2	<a href="https://www.keyence.com/ss/products/microscope/bz-x/study/principle/structure.jsp">https://www.keyence.com/ss/products/microscope/bz-x/study/principle/structure.jsp</a>	
3	<a href="https://www.ncbi.nlm.nih.gov/pmc/articles/PMC6604941/#">https://www.ncbi.nlm.nih.gov/pmc/articles/PMC6604941/#</a>	
4	<a href="https://bio.libretexts.org/@go/page/9188">https://bio.libretexts.org/@go/page/9188</a>	
5	<a href="https://courses.lumenlearning.com/boundless-microbiology/chapter/microbial-nutrition/">https://courses.lumenlearning.com/boundless-microbiology/chapter/microbial-nutrition/</a>	

<b>Methods of Evaluation</b>		
<b>Internal Evaluation</b>	Continuous Internal Assessment Test	25 Marks
	Assignments	
	Seminars	
	Attendance and Class Participation	
<b>External Evaluation</b>	End Semester Examination	75 Marks
	Total	100 Marks
<b>Methods of Assessment</b>		
<b>Recall (K1)</b>	Simple definitions, MCQ, Recall steps, Concept definitions	
<b>Understand/ Comprehend (K2)</b>	MCQ, True/False, Short essays, Concept explanations, Short summary or overview	
<b>Application (K3)</b>	Suggest idea/concept with examples, Suggest formulae, Solve problems, Observe, Explain	
<b>Analyze (K4)</b>	Problem-solving questions, Finish a procedure in many steps, Differentiate between various ideas, Map knowledge	
<b>Evaluate (K5)</b>	Longer essay/ Evaluation essay, Critique or justify with pros and cons	
<b>Create (K6)</b>	Check knowledge in specific or off beat situations, Discussion, Debating or Presentations	

### Mapping with Programme Outcomes:

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11
CO1					M	M				M	
CO2										M	M
CO3											S
CO4				M							S
CO5				M							S

<b>MICROBIOLOGY II</b>			
<b>Course Objectives</b>			
CO1	Learn the Bacterial and Viral diseases.		
CO2	Understand the Fungal and Protozoan diseases.		
CO3	Impart knowledge about the Environmental Microbiology.		
CO4	Acquire knowledge about the Food Microbiology		
CO5	Familiarize the biofertilizers and biopesticides		
<b>UNIT</b>	<b>Details</b>	<b>No.of Hours</b>	<b>Course Objectives</b>
I	Bacteria diseases – Symptoms and control of Respiratory Diseases: <i>S. pyogenes</i> , <i>M. tuberculosis</i> Gastrointestinal Diseases: <i>E. coli</i> , <i>S. typhi</i> , <i>Vibrio cholerae</i> . Fungal diseases - Causative organism, Symptoms and control Cutaneous mycoses: Tinea pedis (Athlete’s foot) Systemic mycoses: Histoplasmosis Opportunistic mycoses: Candidiasis.	8	CO1
II	Viral diseases – Causative organism, Symptoms and control of Polio, Hepatitis, Rabies, Dengue, AIDS, Influenza with brief description of swine flu, Ebola, Chikungunya. Protozoan diseases - Causative organism, Symptoms, mode of transmission and control Malaria, Kala-azar.	8	CO2
III	Microbiology of Air – Sources of air borne organisms, Air borne diseases. Water borne diseases; Purification of water. Sewage treatment – Physical, Chemical and Biological methods.	8	CO3
IV	Sources of contamination and spoilage of foods; Food Preservation; Fermentation products - Bread and Alcoholic beverages (Beer & Wine); Fermented dairy products – Cheese & Yogurt. SCP- <i>Spirulina</i> and Mushroom.	8	CO4
V	Biofertilizers – Definition, Types, Importance and Advantages; Nitrogen fixing microorganisms; Phosphate solubilizing microorganisms; Biopesticides	8	CO5
	Total	40	



<b>Course Outcomes</b>		
<b>Course Outcomes</b>	On completion of this course, students will;	
CO1	Gain Knowledge of common bacterial and fungal diseases.	PO5, PO6, PO10
CO2	Gain Knowledge of common viral and protozoan diseases.	PO10
CO3	Understand the air, water and waste water microbiology	PO11
CO4	Understand the food and dairy microbiology	PO4, PO11
CO5	Utilize the knowledge of biofertilizers and biopesticides. for sustainable agriculture.	PO4, PO11
<b>Text Books</b>		
1	Kanunga R. (2017). Ananthanarayanan and Panicker's Text book of Microbiology. (10 <sup>th</sup> Edition). Universities Press (India) Pvt. Ltd.	
2	Dubey, R.C. and Maheshwari D.K. (2010). A Text Book of Microbiology. S. Chand & Co.	
3	Rajan S. (2007). Medical Microbiology. MJP publisher.	
4	Arora, D. R. and Arora B. B. (2020). Medical Parasitology. (5 <sup>th</sup> Edition). CBS Publishers & Distributors Pvt. Ltd. New Delhi.	
5	Frazier WC and Westhoff DC (2014). Food Microbiology. Tata McGraw Hill Publishing Company Ltd. New Delhi	
6	Subba Rao. N. S. (2017). Soil Microbiology. (5 <sup>th</sup> Edition). MedTech Publishers.	
7	Daniel. C. J. (2006). Environmental Aspects of Microbiology. (2 <sup>nd</sup> Edition). Bright Sun Publications.	
<b>References Books</b>		
1	Salle A. J. (2007). Fundamental Principles of Bacteriology. (4 <sup>th</sup> Edition). Tata McGraw-Hill Publications.	
2	Collee J.C. Duguid J.P. Foraser, A.C, Marimon B.P, (1996). <u>Mackie &amp; McCartney Practical Medical Microbiology</u> . 14 <sup>th</sup> edn, Churchill Livingston.	
3	Pepper I. L., Gerba C. P. and Gentry T. J. (2014). Environmental Microbiology (1 <sup>st</sup> Edition). Academic Press, Elsevier.	
4	Bitton, G. (2011). Wastewater Microbiology. (4 <sup>th</sup> Edition). Wiley-Blackwell.	
5	Madigan M.T., Martinko J.M., Stahl D.A, and Clark D. P. (2010). Brock - Biology of Microorganisms, 13 <sup>th</sup> Edition Benjamin-Cummings Pub Co.	
<b>Web Resources</b>		
1	<a href="http://textbookofbacteriology.net/nd">http://textbookofbacteriology.net/nd</a>	
2	<a href="https://www.adelaide.edu.au/mycology/">https://www.adelaide.edu.au/mycology/</a>	
3	<a href="https://en.wikipedia.org/wiki/Virology">https://en.wikipedia.org/wiki/Virology</a>	
4	<a href="http://www.environmentshumail.blogspot.in/">www.environmentshumail.blogspot.in/</a>	
5	<a href="http://www.fsis.usda.gov/">http://www.fsis.usda.gov/</a>	

<b>Methods of Evaluation</b>		
<b>Internal Evaluation</b>	Continuous Internal Assessment Test	25 Marks
	Assignments	
	Seminars	
	Attendance and Class Participation	
<b>External Evaluation</b>	End Semester Examination	75 Marks
	Total	100 Marks
<b>Methods of Assessment</b>		
<b>Recall (K1)</b>	Simple definitions, MCQ, Recall steps, Concept definitions	
<b>Understand/ Comprehend (K2)</b>	MCQ, True/False, Short essays, Concept explanations, Short summary or overview	
<b>Application (K3)</b>	Suggest idea/concept with examples, Suggest formulae, Solve problems, Observe, Explain	
<b>Analyze (K4)</b>	Problem-solving questions, Finish a procedure in many steps, Differentiate between various ideas, Map knowledge	
<b>Evaluate (K5)</b>	Longer essay/ Evaluation essay, Critique or justify with pros and cons	
<b>Create (K6)</b>	Check knowledge in specific or offbeat situations, Discussion, Debating or Presentations	

### Mapping with Programme Outcomes:

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11
CO1					M	M				M	
CO2										M	M
CO3											S
CO4				M							S
CO5				M							S

<b>MICROBIOLOGY PRACTICAL</b>		
<b>Course Objectives</b>		
CO1	Acquire knowledge on cleaning of glass wares and sterilization.	
CO2	Gain knowledge on media preparation and cultural characteristics.	
CO3	Learn the pure culture technique	
CO4	Learn the microscopic techniques and staining methods.	
CO5	Acquire knowledge to isolate the microorganisms from the environment	
<b>UNIT</b>	<b>Details</b>	
I	Cleaning of glass wares, Microbiological good laboratory practice and safety. Sterilization of glass wares and media.	
II	Media preparation: liquid and solid media. Antibiotic sensitivity testing – Kirby Bauer method	
III	Pure culture techniques: streak plate, Serial dilution – spread plate and pour plate.	
IV	Staining techniques: smear preparation, simple staining and Gram's staining. Motility demonstration – Hanging drop technique. Fungal identification by Lactophenol cotton blue staining technique.	
V	Isolation of microorganisms from air, soil and sewage. Testing the quality of milk - MBRT	
<b>Course Outcomes</b>		
<b>Course Outcomes</b>	On completion of this course, students will;	
CO1	Practice sterilization methods	PO4, PO7, PO8, PO9, PO11
CO2	Learn to prepare different media and their quality control.	PO4, PO7, PO8, PO9
CO3	Learn streak plate, pour plate and serial dilution and pigment production of microbes.	PO4, PO7, PO8, PO9, PO11
CO4	Understand Microscopy methods, different Staining techniques and motility test.	PO4, PO7, PO8, PO9
CO5	Acquire knowledge to isolate bacteria from the environment	PO4, PO7, PO8, PO9
<b>Text Books</b>		
1	James G Cappucino and N. Sherman MB (1996). A lab manual Benjamin Cummins, New York 1996.	
2	Kannan. N (1996). Laboratory manual in General Microbiology. Palani Publications.	
3	Sundararaj T (2005). Microbiology Lab Manual (1 <sup>st</sup> edition) publications.	
4	Gunasekaran, P. (1996). Laboratory manual in Microbiology. New Age International Ld., Publishers, New Delhi.	
5	R C Dubey and D K Maheswari (2002). Practical Microbiology. S. Chand Publishing.	
<b>References Books</b>		

1	Atlas. R (1997). Principles of Microbiology, 2 <sup>nd</sup> Edition, WM.C.Brown publishers.
2	Amita J, Jyotsna A and Vimala V (2018). Microbiology Practical Manual. (1 <sup>st</sup> Edition). Elsevier India
3	Talib VH (2019). Handbook Medical Laboratory Technology. (2 <sup>nd</sup> Edition). CBS
4	Wheelis M, (2010). Principles of Modern Microbiology, 1st Edition. Jones and Bartlett Publication.
5	Lim D. (1998). Microbiology, 2 <sup>nd</sup> Edition, WCB McGraw Hill Publications.
<b>Web Resources</b>	
1	<a href="http://www.biologydiscussion.com/micro-biology/sterilisation-and-disinfection-methods-and-principles-microbiology/24403">http://www.biologydiscussion.com/micro-biology/sterilisation-and-disinfection-methods-and-principles-microbiology/24403</a> .
2	<a href="https://www.ebooks.cambridge.org/ebook.jsf?bid=CBO9781139170635">https://www.ebooks.cambridge.org/ebook.jsf?bid=CBO9781139170635</a>
3	<a href="https://www.grsmu.by/files/file/university/cafedry//files/essential_microbiology.pdf">https://www.grsmu.by/files/file/university/cafedry//files/essential_microbiology.pdf</a>
4	<a href="https://microbiologyinfo.com/top-and-best-microbiology-books/">https://microbiologyinfo.com/top-and-best-microbiology-books/</a>
5	<a href="https://www.cliffsnotes.com/studyguides/biology/microbiology/introduction-to-microbiology/a-brief-history-of-microbiology">https://www.cliffsnotes.com/studyguides/biology/microbiology/introduction-to-microbiology/a-brief-history-of-microbiology</a>

**Mapping with Programme Outcomes:**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11
CO1				M			L	M	L		M
CO2				S			L	L	L		
CO3				S			M	M	L		M
CO4				S			M	L	L		
CO5				S			M	L	L		

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